

EXHIBIT A

CHEMICAL ANALYSIS

A SERIES OF MONOGRAPHS ON ANALYTICAL
CHEMISTRY AND ITS APPLICATIONS

Editors

J. D. WINEFORDNER

Editor Emeritus I. M. KOLTHOFF

VOLUME 117



A WILEY-INTERSCIENCE PUBLICATION

JOHN WILEY & SONS, INC.

New York / Chichester / Brisbane / Toronto / Singapore

Applications of Fluorescence in Immunoassays

ILKKA A. HEMMILÄ

Department of Biochemistry
University of Turku
Turku, Finland



1991

A WILEY-INTERSCIENCE PUBLICATION

JOHN WILEY & SONS, INC.

New York / Chichester / Brisbane / Toronto / Singapore

SAO 127/001
BOEHRINGER MANNHEIM GMBH
Forschungszentrum Tübingen
- Bibliothek -

NR. 328

5. 32

ROCHE PZ ABT. LR-IRC 1

ROCHE PZ ABT. LR-IRC

15/04 '02 MO 14:25 [SE/EM NR 9253]

9. APR. 2002 15:53

PREFACE

Fluorescence provides a diversified and sensitive detection system applied in the versatile field of immunological techniques. The application of antibodies labeled with fluorescent probes dates back to the 1940s, when Coons et al. introduced the microscopic immunofluorescence staining technique. During the 1970s fluorescence was considered as a promising and potentially very sensitive detection system in the search for alternative labels to replace radioisotopic tracers in immunoassays. Regardless of the number of assays developed and also successfully applied in certain areas, the inherent vulnerability of fluorescence detection to background interferences hindered its application in assays requiring high sensitivity. The recent development of fluorescence instruments, assay technologies and fluorescent probes has, however, resulted in assay techniques producing one of the highest available sensitivities, and fluorometric immunoassays also offer real alternatives to the sensitive radioisotopic immunoassays. The ability of fluorometric detection to combine spectral, temporal and spatial resolution offers a powerful tool for future immunoassay development, too.

The present monograph describes the basic prerequisites for a fluorometric immunoassay; the antibody, the immunological technology, the fluorescent probe and the instrument, as well as gives a profile of the clinical applications of the various assay technologies.

The author would like to express his gratitude to Mrs. Aini Taivonen for recording the fluorescence spectra included in the monograph, Mr. Raimo Harja, M.Sc., for updating the authors knowledge about lasers and detectors, and to Mr. Pertti Hurskainen, M.Sc., for proofreading the chapter discussing DNA-based assays.

ILKKA A. HEMMILÄ

Bibliothek Eingang: 2.6.93
ACC. Nr. 245803

In recognition of the importance of preserving what has been written, it is a policy of John Wiley & Sons, Inc., to have books of enduring value published in the United States printed on acid-free paper, and we exert our best efforts to that end.

Copyright © 1991 by John Wiley & Sons, Inc.

All rights reserved. Published simultaneously in Canada.

Reproduction or translation of any part of this work beyond that permitted by Section 107 or 108 of the 1976 United States Copyright Act without the permission of the copyright owner is unlawful. Requests for permission or further information should be addressed to the Permissions Department, John Wiley & Sons, Inc.

Library of Congress Cataloging in Publication Data:

Hemmilä, Ilkka A.

Applications of fluorescence in immunoassays / Ilkka A. Hemmilä.

1. cm. — (Chemical analysis, ISSN 0069-2883 ; v. 117)

"A Wiley-Interscience publication."

Includes bibliographical references and index.

ISBN 0-471-51091-2

1. Immunoassays. 2. Fluorescent antibody technique.

3. Fluorescent antigen technique. I. Title. II. Series.

QR187.14H46 1991

616.0756—dc20

90-27188

CIP

Printed in the United States of America

10 9 8 7 6 5 4 3 2 1

ROCHE PZ RBT, LR-IRC
ROCHE PZ RBT, LR-IRC

15/04 '02 MO 14:25 [SE/EM NR 9253]

NR. 328 5.33

9. APR. 2002 15:54

the inherent nitrogen as a marker. The use of fluorescent compounds as sensitive marker substances coupled to antibodies was invented by Albert Coons and his colleagues in the early '40s, when they developed immunofluorescence staining techniques for microbes (5, 6).

The study of antibody production in diabetic patients treated with insulin led to the development of the radioimmunoassay in the late '50s by Berson and Yalow (7, 8); this method has had a major impact on the acceptance of immunological techniques in the field of routine clinical diagnosis.

In their early days radioimmunoassays were exclusively applied for determinations of peptide hormones. Since the pioneering work of Landsberger in 1946 (9), antibodies have also been produced for small (molecular weight under 10,000) compounds called haptens, for compounds which as such are unable to elicit antibody production but must be bound to larger carrier molecules to form immunogenic conjugates. The production of antisera against haptenic molecules, such as steroids (10) or thyroid hormones (11), opened a new dimension for immunoassays. Since then antibodies have been produced against an enormous number of antigens and biological and synthetic compounds, and these have been applied in a variety of ways for analyzing these compounds. Modern biotechnology has revolutionized antibody production, and genetic engineering opens totally new perspectives for their future applications.

2.1. IMMUNOGENIC RESPONSE

An antigen is an immunogenic compound which can elicit a strong immune response in an immunized animal. An immunogenic antigen can be a peptide, protein, polysaccharide, polynucleotide, or almost any polymeric compound containing functional groups on its surface recognized by antibody-producing B-lymphocytes. The primary recognition by the membrane bound receptor proteins of lymphocytes triggers the complex process of maturation of antibody producing B-cells and the subsequent production of large quantities of antibodies.

The production of antisera of high titer, affinity, and specificity requires substantial amounts of chemically pure antigens. A large amount is needed for repeated immunizations of test animals. High purity is an absolute necessity in order to obviate cross-reactivities with unrelated compounds. The purification and stability problems with some biological compounds can be a limiting factor in antiserum production, but these have been partly overcome with the development of methods for producing monoclonal antibodies (Chapter 2.2).

2.1.1. Haptenic Antigens

Haptenic antigens are compounds which because of their small size cannot elicit immunoresponse. Generally the molecular weight limit for immunogenic response is around 10,000. Because of the difficulties in producing anti-hapten antibodies, the first real immunoassays were developed for peptides or proteins, and actually the first "specific protein binding assays" of haptenic molecules, developed by Roger Elkins et al. in the early '60s, are naturally occurring specific binding proteins, thyroxine binding globulin for labeled thyroxine (12) and intrinsic factor for labeled B₁₂-vitamin (13).

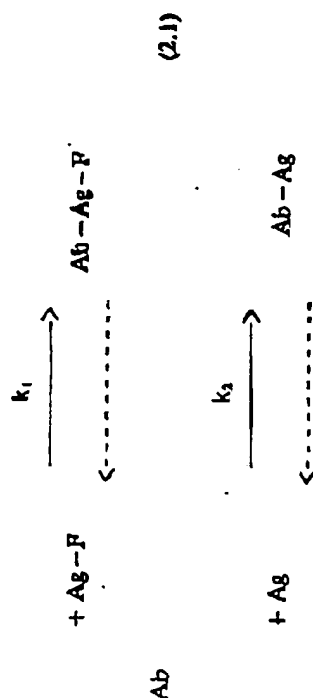
The production of anti-hapten antibodies was invented in the late '40s (9), and anti-steroid antibodies were produced in 1957 (10). It was several years, however, before these were applied for making radioimmunoassays. For eliciting immunoresponse the haptenic molecules need first to be coupled to a suitable carrier. Bovine serum albumin is the most often used carrier protein for immunizations, mainly because of its solubility and availability. Other proteins, like keyhole limpet hemocyanin, have been preferred later on because of their high immunogenicity and coincident contribution of the production of anti-hapten antibodies with high titer and affinity (14).

The production of anti-hapten antibodies of predetermined specificity is often problematic, partly because the coupling of the compound to a carrier can block important epitopic sites needed for specificity and partly because of recognition of the linking arm between the hapten and carrier by the produced antibodies. Since the antibodies are able to bind structures equal to about 7 amino acid residues (15), an anti-hapten antibody most often recognizes simultaneously part of the linking group and spacer arm used in conjugation reaction for immunization (bridge recognition).

Bridge recognition is especially problematic for steroid immunoassays (16, 17) and is encountered when labeled steroids (tracers) or immobilized steroids (e.g., solid-phase reagent) are prepared using the same position of the steroid (site homology) or same linking arm (bridge homology) as used for preparing the immunogenic conjugate. With such conjugates the competitive binding between the limited amount of antibody, labeled antigen, and the unknown amount of sample antigen (or standard) favors the reaction between tracer and antibody with poor replacement; the rate constant k_r is much higher than k_d (Eq. 2.1). The poor replacement results in insufficient slope to the standard curve and low assay sensitivity because the sample antigen is unable to compete with the tracer in binding to antibodies.

Accordingly, the production of immunogenic conjugates for steroid immunization is better performed after selecting different spacer arms or sometimes even different positions for attachment on the steroid structure (Fig. 2.1).

ANTIBODIES AS ANALYTICAL REAGENTS



For example, considerably higher sensitivity was obtained in an assay of 17-hydroxyprogesterone when using a bridge heterologous tracer as compared to a respective homologous system (18). Similarly, the equilibrium time required for ligand displacement in an assay of estradiol shortened from 10 h to 1 min when changing from a homologous system to heterologous (19). The requirement of site homology depends greatly on the analyte and antibodies used. In BIA of cortisol Arakawa et al. (20) used cortisol-6 α -hemisuccinate for producing the antigen conjugate for solid-phase immobilization and 3-carboxymethylcholone conjugate for producing the marker-enzyme tracer. On the other hand, in the experiments of Kobayashi et al. (21) and Mikola and Miettinen (22), cortisol could be assayed only with a site homologous system. Tietz and Andrich (23) tested spacer arms with a site homologous system. They found that a reasonably long spacer was an absolute necessity and that the chemical structure of the spacer may also have a major effect on bridge recognition.

2.2. MONOCLONAL ANTIBODIES

In 1975 Köhler and Milstein (24) made the first monoclonal antibodies of predetermined specificity by fusing a spleen cell line producing the specific antibodies with a myeloma cell line capable of continuous growth in cell culture. Since then the advent of monoclonal antibodies has had an enormous impact on many fields of biomedical research (25, 26). It was soon realized that the technique would revolutionize the immunoassay field as well, and it has raised great expectations also in immunotherapy, imaging, and biotechnology.

Monoclonal antibodies are rapidly gaining a dominant position in immunoassays, especially from a commercial point of view, because of their unlimited supply, molecular homogeneity, and defined, unchanged properties. The production and use of monoclonal antibodies has also been

MONOCLONAL ANTIBODIES

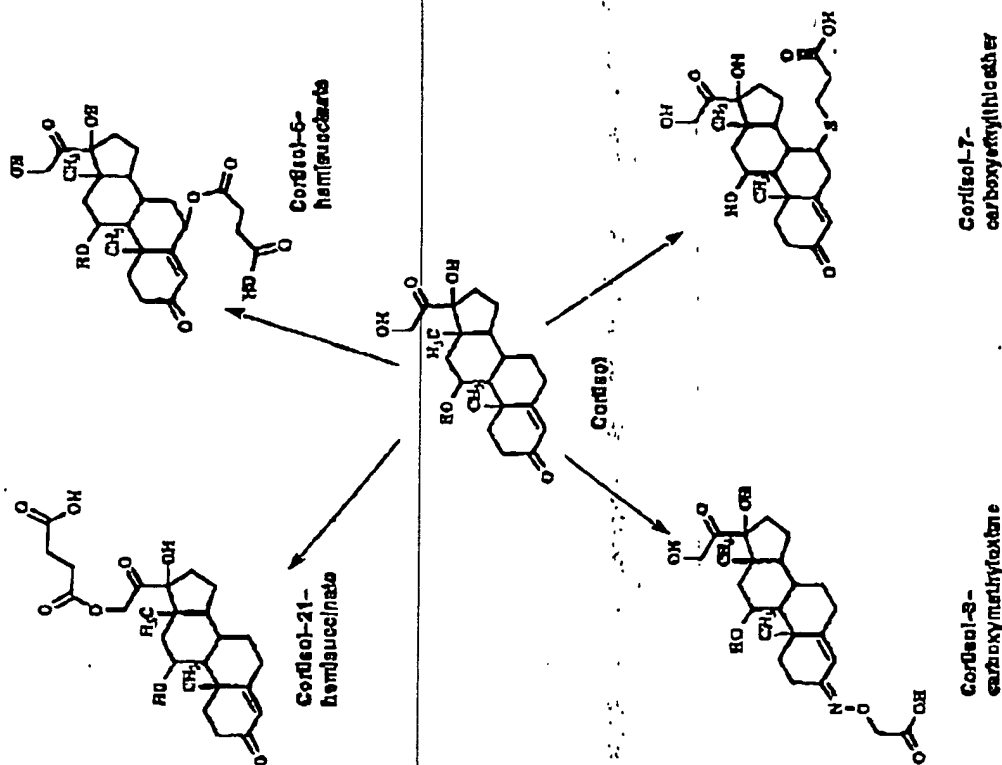


Fig. 2.1. Site heterologous routes to prepare cortisol derivatives for immunization and for labeling.

and it was about 20 years before FPIA reached routine clinical laboratories—when Abbott adopted the technology and developed clinical instruments and numerous kits based on FPIA technology.

Today numerous homogeneous assay principles have been introduced. Several technologies are commercialized and have found quite extensive applications in certain areas, especially in measuring drugs (therapeutic drug monitoring, TDM, and tests for illicit drugs).

8.3.1. Fluorescence Polarization Immunoassays

The efficiency of light absorption by a fluorophore is dependent on the angle between the electronic dipole of the exciting light and the absorption oscillators of the molecule. A polarized light will excite only those molecules that have their absorption oscillators parallel to the plane of exciting light. The polarization level of the resultant emission depends on the lifetime of the excited state (τ) and the rotational motion of the molecule. For steady-state measurement polarization is generally expressed by the Perrin equation (Eq. 8.1) (1982).

$$(1/p - 1/3) = (1/p_0 - 1/3) (1 + 3\tau/p) \quad (8.1)$$

The rotational relaxation time, p , can be calculated for a spherical molecule according to Equation 8.2.

$$p = 3 \eta v / kT \quad (8.2)$$

Rotational relaxation time is directly proportional to the volume of the molecule (size and shape) and viscosity (η) of the medium.

A large molecule, such as an antibody, has a tumbling time typically around 10 to 100 ns, whereas small molecules, such as haptens, have tumbling times around 0.1 to 1 ns. In steady-state polarization measurement (continuous excitation with polarized light), the resulting polarization of emission depends on the size and shape of the labeled substance and the ratio of rotational relaxation time to the decay time of the fluorochrome. This forms the basis for measuring binary binding reactions—for example, in immunoreaction.

To be practical for an immunoassay, the change in molecular volumes during the immunoreaction needs to be high enough, such as it is during the binding of the haptenic tracer to its antibodies. It gives a practical limit for the size of antigen, which should be below 20,000. The decay time of the fluorophore needs to be longer than the rotational time of the haptenic tracer but shorter than the rotational time of the formed complex. Fluorescein (τ

4.5 n) accordingly works very well for normal FPIA, the polarization of which increases dramatically upon the binding of the fluorescein-labeled hapten to the respective antibodies.

For large antigens, fluorochromes with somewhat longer decay times have been tested. With proteins the intramolecular tumbling becomes problematic, however, when using steady-state measuring (1983). So far no applications have been made using large binding entities (e.g., microbeads) and long τ probes.

In addition to the size limitations of FPIA, problems arise also from the low affinity nonspecific binding properties of serum proteins, especially that of albumin, which increases the polarization level nonspecifically. To avoid the albumin effect, a simple dilution jump has been used (1984), or various sample pretreatments are required. In the pretreatment solutions either chaotropic ions, proteolytic enzymes, protein precipitating reagents, or solvents are used. The pretreatment of samples is especially needed for analytes that require a high sensitivity that does not allow for high dilutions.

The principle of fluorescence polarization was developed by Perrin in 1926 (1982). About 30 years later, in 1952, the technology was applied in biological systems by Weber (1985). For monitoring immunoreactions, the fluorescence polarization technique has been used since 1961, since the pioneering work of Dandliker et al. (277), who studied the interaction of fluorescein-labeled penicillin (279), ovalbumin (280), and estrone (1986) with their specific binding proteins or receptors. Dandliker has also written a number of review articles about the principle and applications of fluorescence polarization (282, 1984, 1987).

The experimental studies of FPIA during the 1960s and 1970s were conducted with research fluorimeters equipped with polarization accessories and have resulted in a limited number of clinical applications (Table 8.9), mainly because of the lack of appropriate instruments for routine assays.

Table 8.9. Early Applications of Fluorescence Polarization in Protein Binding Assays

Analyte	Tracer	Assay type	Reference
Anti-penicillin-Ab	FITC-Penicillin	Direct	279
Estrone receptor	FITC-Estrone	Direct	1986
Anti-ovalbumin-Ab	FITC-Ovalbumin	Direct	280
Anti-Conalbumin-Ab	FITC-Conalbumin	Direct	281
Trypsin	FITC-Casein	Direct	283
Anti-insulin-Ab	FITC-Insulin	Direct	283
hCG	FITC-hCG	Competitive	282, 1988
Penicillin	FITC-Ag. #	Competitive	1989

Table 8.10. Applications of TD₂ (continued)

Analyte	Reference	Evaluation
Hormones		
Free cortisol/10		1258
Estriol	1259	
OH-Indole acetal	1260	
Proteins and peptides		
Angiotensin	1261	
CRP	1262	
Transferrin		1263, 1264, 1265

Wider usage of competitive FPIA in clinical routine started during the early 1980s, when Abbott introduced an automated instrument designed for clinical FPIA applications (455, 456, 1190). With instruments of various stages of automation (TD₂, AD₂ and IM₂) and over 50 different kits, FPIA has become one of the most used FIA in clinical chemistry (379, 1191). Lately Abbott has been accompanied by other reagent and instrument manufacturers, such as Roche Diagnostics, CANAM Diagnostics, Colony, Sankyo, Lundbeck of Oregon Inc. (RNNOFLUORIM FPIA), Polymed Co. and Source Scientific Systems (FocusTM FPIA fluorometer).

The reagent pack of TD₂ generally contains a pretreatment solution, and serum, and antigen labeled with a fluorescent derivative. The instrument performs the required dilutions, records the blank value to be subtracted, and measures the final polarization level. The technology is used primarily in TDM and screening for illicit drugs, but it is also used for some hormones and even for a few proteins, such as globulins, transferrin, and CRP. Table 8.10 summarizes examples of the articles describing FPIA applications performed on TD₂, AD₂, or the automated IM₂, including the numerous evaluations of the existing kits and other FPIA applications of the TD₂ instrument. Some of the assays, such as the assay of cyclosporine, have spanned a great number of evaluations, partly collected in the table.

The research group of Prof. Landon has developed FPIAs since 1976 (1266) and has developed analytical applications for the determination of hormones and drugs. They have been able to simplify the technology further by using a one-step, one-reagent method based on antibodies pre-equilibrated with FITC-labeled antigens. By adjusting the respective affinities so that ligand displacement can take place rapidly, this LJDIA principle provides an extremely simple and rapid analysis. Assays are performed with various research fluorometers, including the Perkin Elmer LS 20 Polarization Fluorometer particularly developed for clinical routine assays (457). FPIA applications performed with homemade reagents are listed in Table 8.11

Table 8.10. Applications of TD₂

Analyte	Reference	Evaluation
Antibiotics		
Gentamicin	1192	1193-1196
Tetracycline	1192	
Amikacin	1192	1197
Spectinomycin	1198	
Nerfthacin		1195, 1199
Isopropylid		1200
Vencomycin		1201
Aztreonam		1202
Anticonvulsants		
Phenytoin	365, 1203	1194, 1204, 1205
Phenobarbital	365, 1203	1194, 1204
Carbamazepine		1204, 1206, 1207
Valproic acid		1204, 1208
Antiarrhythmics		
Quinidine/free		1209, 1210
Hydroquinidine		1211
Lidocaine		1212
Disopyramide/free	1213	
MEGX	1214	
Other drugs		
Theophylline	1215, 1216	1194, 1217, 1218
Digoxin		1194, 1219-1221
Methyldigoxin	1222	
Benzodiazepine	1223	
Flecainamide		1224, 1225
Paracetamol		1226
Methotrexate		1227
Cyclosporine	1228-1231	1228-1243
Other drugs		
Morphine	1244	
Amphetamine	1245	
MHPG	560	1246
Barbiturates		1247
Opiates		1247
Cocaine metabol.	1248	1249-1251
Hormones		
Thyroxine (T ₄)	1252	1253, 1254
FT ₄		1253
T ₄	1255	
T uptake	1256	
Cortisol		1257

Table 8.11. Research Applications of FPIA

Analyte	Label	Reference
TDM		
Codeine	FITC	1255
Phenylephrine	FITC	372
Phenytoin	2-Naphthol-sulfonamide	390
Valproic acid	FITC	1257
Paracetamol	FITC	457, 1253
Quinine	FITC	1259
Theophylline	FITC	489
Theophylline	Umbelliferyl	672
Salicylate	FITC	1270
Illicit drugs		
Opiates	FITC	345
Amphetamine	FITC	456, 1274
Mephentermine	FITC	344, 458
Benzylisopropylamine	FITC	450
Barbiturates	FITC	1272-1274
Valiylisovalerate	FITC	1275
Acidobenzamide	FITC	1276
Hormones		
Cardiol	FITC	370, 373, 1277
Biotin	FITC	1278
Neopterin	FITC	1278
Deoxycurtinol	FITC	1279
Isotriol	Fluorescein	19
Biotin	Lucifer Yellow	588
Testosterone	Lucifer Yellow	588

New manufacturers producing FPIA kits have recently emerged. The kits are intended to be measured either with the existing Abbott TD_x system or with the manufacturer's own instrument, such as the Roche FPIA, developed for the company's Cobas Bio and Cobas Para chemistry analyzers. At the moment, alternative products are concentrated in drug monitoring (Table 8.12).

Relatively little effort has been used to develop FPIAs for larger molecules such as proteins. The problems with proteins are related to their large size and their flexibility, as well as the lower sensitivity and more narrow dynamic range obtainable. TD_x has, however, been applied to some proteins, such as globulins, ferritin, and CRP (Table 8.10) and to analytes that do not require high sensitivities.

Table 8.12. Alternative Commercial FPIA Assays

Analyte	Company	Reference
Codeine	Roche Diag.	1280
Tobramycin	IBC, Innotron Diag.	1281
	Roche Diag.	1280
Vancomycin	IBC, Innotron Diag.	1281
Phenylephrine	Roche Diag.	565
	Roche Diag.	566, 1282
Phenobarbital	IBC, Innotron Diag.	1281
	Roche Diag.	566
Carbamazepine	IBC, Innotron Diag.	1281
Theophylline	Roche Diag.	566
	CANAM	1280, 1283
	IBC, Innotron Diag.	1281
	Colony	1284
Quinine	Roche Diag.	566, 1282, 1285
Practolone	Roche Diag.	1286
Digoxin	Roche Diag.	1287
Procainamide	Roche Diag.	566, 1282
NAPA	Roche Diag.	566
Dilantin	Roche Diag.	1280

Ufios et al. (1188) have made an FPIA for urine hCG using FITC-labeled hCG as the tracer. Reportedly they observed a rather wide dynamic range in the assay—from 0.27 to 64 µg/ml. Yamaguchi et al. (501) were able to measure insulin with a competitive FPIA with a dynamic range from 40 to 600 mU/ml, but only from pure insulin preparations and not from serum samples. A similar insulin FPIA has also been tested by Nishipathorn and McGown (1288), who studied the fluorescence intensity changes, decay-line changes, and polarization changes of FITC-labeled insulin during the immunoreaction. Assays of smaller peptides, such as angiotensin (1261) and neocortisol (454), can be developed more easily.

One approach toward protein FPIAs has been the development of fluorescent probes with longer decay times. Dansyl and umbelliferone derivatives have been tested for FPIA of CK-MB (617). The longer decay-time emitter, dansyl derivative, proved to be too flexible to give an acceptable polarization level. Ufios and Cattanova used Lucifer Yellow as a label in a direct assay of IgM (1289). The assay was based on a smaller binding unit, the Fab-fragment of a monoclonal antibody, labeled with the fluorophore and used as a direct reagent for the larger antigen, IgM.